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Efficient Iodine-catalyzed Synthesis of 3-Aryl 4(3H)-quinazolinones

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Compounds with a 4-quinazolinone motif have displayed a variety of interesting pharmacological properties such as antimalarial, antitumor, anticonvulsant, antiinflammatory, fungicidal, antimicrobial and calcilytic activities.¹⁻⁷ An examination of the literature indicates that 4-quinazolinones substituted at the 3-position are attracting the attention of chemists because they appear to have important sedative, hypnotic and anticonvulsive effects.⁸ There are several methods for the preparation of this class of compounds. The most direct procedure includes the condensation of anthranilic acid, ortho esters and amines in the presence of various catalysts such as NaHSO₄ or Amberlyst-15,⁹ Yb(III)-resin,¹⁰ Yb(OTf)₃,¹¹ Bi(TFA)₃-[nbp]FeCl₄ ionic liquid,¹² La(NO₃)₃·6H₂O or *p*-toluenesulfonic acid,¹³ Keggin-type heteropolyacid,¹⁴ SnCl₄·4H₂O,¹⁵ and SiO₂-FeCl₃.¹⁶ While this type of multi-component reaction (MCR) can produce the desired products in a single step, some of these methods have certain drawbacks such as expensive catalysts, high temperatures (60-80°C), long reaction times (20 h), and the use of harmful organic solvents. Most so-called "solvent-free" techniques still require an organic solvent such as CH_2Cl_2 during work up to remove the support, reagents or side-products. Therefore, it is desirable to develop simple, efficient, and green methods for the synthesis of 4(3H)-quinazolinones.

In this regard, in recent years, molecular iodine has received considerable attention as an inexpensive and easily available catalyst for various organic transformations such as esterification,¹⁷ oxidation,¹⁸ acetylation,¹⁹ Hantzsch,²⁰ Biginelli reaction,²¹ aziridination,²² allylation,²³ Michael addition,²⁴ Johnson–Claisen rearrangement,²⁵ and Diels–Alder reactions.²⁶ We report the synthesis of 3-aryl 4(3*H*)-quinazolinones by the condensation of anthranilic acid **1**, ortho esters **2** and aromatic amines **3** in the presence of a catalytic amount of I₂ at room temperature without solvent (*Scheme 1*).

A model reaction of anthranilic acid with triethyl orthoformate and aniline carried out at room temperature in the absence of catalyst gave only a 5% yield after 1 h. However, a

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quantitative yield (99%) of the product was obtained after only 40 min when 3 mol% of iodine was added. It was observed that 3 mol% of iodine catalyzed the reaction efficiently but increasing the amount of catalyst did not lead to better yields.

Table 1 illustrates the scope and generality of the reaction of anthranilic acid **1**, triethyl or trimethyl orthoformate **2**, and different aryl amines. Aryl amines carrying either electrondonating or electron-withdrawing groups reacted smoothly at room temperature and, except in the case of 2-nitroaniline, steric effects did not influence the yields significantly. Although benzylamine gave a moderate yield of the desired product (Entry **4n**), no product was

Product	R_1	Time ^b (time) ^c (min)	Yield ^b (yield) ^c (%)	mp (°C)	
				Found	<i>lit</i> . ¹¹
4a	Ph	40 (30)	99 (92)	140–141	139–140
4b	2-MeC ₆ H ₄	90 (120)	99 (99)	158-159	
4c	3-MeC ₆ H ₄	180 (360)	99 (60)	137–139	136–137
4d	4-MeC ₆ H ₄	18 (42)	97 (87)	143–146	146–147
4e	2-MeOC ₆ H ₄	40 (90)	99 (99)	151-153	
4f	4-MeOC ₆ H ₄	18 (30)	99 (92)	132-134	
4g	$2-ClC_6H_4$	5 (10)	86 (78)	117-120	
4h	$4-ClC_6H_4$	9 (120)	81 (90)	122-124	
4i	$4-BrC_6H_4$	6 (12)	98 (82)	149–151	
4j	$2-NO_2C_6H_4$	330 (150)	41 (45)	153-155	156–158
4k	$3-NO_2C_6H_4$	5 (2)	82 (93)	150-153	154–156
41	$4-NO_2C_6H_4$	5 (2)	99 (99)	166–168	165–166
4m	$4-HO_2CC_6H_4$	1(1)	87 (86)	240-242	
4n	PhCH ₂	2 (30)	30 (61)	154–155	_
40	$n-C_4H_9$	1200 (1200)	0 (0)	—	—

Table 1	
I ₂ -Catalyzed Synthesis of 3-Substituted $4(3H)$ -Quinazolinones $4(a-4)$	o) ^a

^a) The structures of the products were determined from spectral (IR and ¹H NMR) data and elemental analysis.

^b) Time and yield with triethyl orthoformate.

^c) Time and yield with trimethyl orthoformate.

obtained from *n*-butylamine (Entry **40**). All crude products after simple washing were pure enough (testified by elemental analysis) without further purification.

In summary, we disclose here an efficient method for multi-component synthesis of 4(3H)-quinazolinones under mild conditions, with simple work-up, and using ethanol as the only solvent.

Experimental Section

Melting points were determined using an RY-1 micromelting point apparatus and were uncorrected. Infrared spectra were recorded on a Scimitar 2000 series Fourier Transform instrument of VARIAN. ¹H NMR spectra were recorded on a Bruker ARX-500 spectrometer in DMSO- d_6 using TMS as an internal standard. Elemental analyses were carried out on an EA 2400II elemental analyzer (Perkin-Elmer).

Representative Procedure for the Synthesis of 4(a-o)

A mixture of anthranilic acid (1.371 g, 10 mmol), the orthoester (12 mmol), the amine (12 mmol), and I₂ (0.076 g, 0.3 mmol) was stirred at room temperature for an appropriate time (*Table 1*). The reaction was monitored by TLC (silica gel 60 F 254 TLC plates, cyclohexane: ethyl acetate, 1: 1 v/v). After completion of the reactions, all products were collected and washed thoroughly with EtOH during filtration and dried. All crude products thus obtained were sufficiently pure (single spot on TLC) without further purification. The obtained products were identified by IR, ¹H NMR, and elemental analysis. Spectral data and elemental analysis for new compounds are as follows:

3-(2-Methylphenyl)quinazolin-4(3*H***)-one (4b)**, white solid. IR (KBr): 1688, 1595, 1489 cm⁻¹; ¹H NMR (500 MHz, DMSO- d_6): δ 8.55 (s, 1H), 8.31 (d, J = 7.2 Hz, 1H), 7.76–7.52 (m, 2H), 7.24–7.07 (m, 5H), 2.31 (s, 3H).

Anal. Calcd. for C₁₅H₁₂N₂O: C, 76.26; H, 5.12. Found: C, 76.38; H, 5.09.

3-(2-Methoxyphenyl)quinazolin-4(3H)-one (4e), white solid. IR (KBr): 1682, 1595, 1456 cm⁻¹; ¹H NMR (500 MHz, DMSO- d_6): δ 8.36 (s, 1H), 8.19 (d, J = 7.7 Hz, 1H), 7.45 (d, J = 6.8 Hz, 1H), 7.25 (t, J = 7.0 Hz, 1H), 7.08–7.03 (m, 2H), 6.94 (t, J = 7.1 Hz, 1H), 6.78 (d, J = 8.2 Hz, 1H), 6.54 (t, J = 7.3 Hz, 1H), 3.85 (s, 3H).

Anal. Calcd. for C₁₅H₁₂N₂O₂: C, 71.42; H, 4.80. Found: C, 71.31; H, 4.83.

3-(4-Methoxyphenyl)quinazolin-4(3H)-one (4f), white solid. IR (KBr): 1717, 1591, 1454 cm⁻¹; ¹H NMR (500 MHz, DMSO-*d*₆): δ 8.54 (s, 1H), 8.05 (d, *J* = 7.5 Hz, 1H), 7.54–7.51 (m, 2H), 7.16 (t, *J* = 7.5 Hz, 1H), 6.98 (dd, *J* = 8.5, 9.5 Hz, 4H), 3.74 (s, 3H). *Anal.* Calcd. for C₁₅H₁₂N₂O₂: C, 71.42; H, 4.80. Found: C, 71.30; H, 4.84.

3-(2-Chlorophenyl)quinazolin-4(3*H***)-one (4g),** yellow solid. IR (KBr): 1667, 1601, 1415 cm⁻¹; ¹H NMR (500 MHz, DMSO- d_6): δ 8.57 (s, 1H), 8.01 (d, J = 7.7 Hz, 1H),

7.73–7.71 (m, 2H), 7.61 (t, J = 7.6 Hz, 1H), 7.25–7.16 (m, 2H), 6.76 (d, J = 8.1 Hz, 1H), 6.54 (t, J = 7.2 Hz, 1H).

Anal. Calcd. for C₁₄H₉ClN₂O: C, 65.51; H, 3.53; N, 10.91. Found: C, 65.63; H, 3.50; N, 10.88.

3-(4-Chlorophenyl)quinazolin-4(3*H***)-one (4h),** pale yellow solid. IR (KBr): 1671, 1616, 1484 cm⁻¹; ¹H NMR (500 MHz, DMSO- d_6): δ 8.53 (s, 1H), 7.71–7.69 (m, 2H), 7.24–7.19 (m, 2H), 6.75 (d, J = 7.5 Hz, 2H), 6.52 (d, J = 7.2 Hz, 2H).

Anal. Calcd. for C₁₄H₉ClN₂O: C, 65.51; H, 3.53; N, 10.91. Found: C, 65.62; H, 3.51; N, 10.86.

3-(4-Bromophenyl)quinazolin-4(3*H***)-one (4i),** white solid. IR (KBr): 1712, 1587, 1443 cm⁻¹; ¹H NMR (500 MHz, DMSO-*d*₆): δ 8.55 (s, 1H), 8.01 (d, *J* = 7.7 Hz, 1H), 7.60 (t, *J* = 7.5 Hz, 1H), 7.46–7.43 (m, 4H), 7.19 (d, *J* = 7.5 Hz, 2H).

Anal. Calcd. for C₁₄H₉BrN₂O: C, 55.84; H, 3.01; N, 9.30. Found: C, 55.73; H, 3.00; N, 9.36.

3-(4-Carboxylphenyl)quinazolin-4(3*H***)-one (4m),** white solid. IR (KBr): 1701, 1592, 1483 cm⁻¹; ¹H NMR (500 MHz, DMSO-*d*₆): δ 10.49 (s, 1H), 8.34 (s, 1H), 7.92–7.88 (m, 3H), 7.71 (t, *J* = 8.5 Hz, 2H), 7.63 (d, *J* = 5.5 Hz, 1H), 7.31 (d, *J* = 8.5 Hz, 1H), 6.56 (d, *J* = 5.5 Hz, 1H).

Anal. Calcd. for C₁₅H₁₀N₂O₃: C, 67.67; H, 3.79. Found: C, 67.77; H, 3.76.

3-Benzylquinazolin-4(3H)-one (4n), white solid. IR (KBr): 1678, 1621, 1459 cm⁻¹;
¹H NMR (500 MHz, DMSO-*d*₆): δ 8.52 (s, 1H), 8.00 (d, *J* = 7.5 Hz, 1H), 7.54–7.47 (m, 2H), 7.41–7.37 (m, 4H), 7.31 (t, *J* = 6.8 Hz, 1H), 7.14 (t, *J* = 7.0 Hz, 1H), 4.61 (s, 2H). *Anal.* Calcd. for C₁₅H₁₂N₂O: C, 76.26; H, 5.12. Found: C, 76.39; H, 5.08.

References

- Y. Takaya, T. Chiba, M. Tanitsu, K. Murata, H. S. Kim, Y. Wataya and Y. Oshima, *Parasitol. Int.*, 47(Suppl.), 380 (1998).
- S. L. Cao, Y. P. Feng, Y. Y. Jiang, S. Y. Liu, G. Y. Ding and R. T. Li, *Bioorg. Med. Chem. Lett.*, 15, 1915 (2005).
- 3. M. J. Kornet, T. Varia and W. Beaven. J. Heterocyclic Chem., 20, 1553 (1983).
- M. R. Yadav, S. T. Shirude, A. Parmar, R. Balaraman and R. Giridhar, *Chem. Heterocycl. Compd.*, 42, 1038 (2006).
- 5. Y. G. Hu, S. J. Yang and M. W. Ding, Phosphorus Sulfur Silicon, 179, 1933 (2004).
- 6. S. Bahadur and M. Saxena, Arch. Pharm. (Weinheim), 316, 964 (1983).
- I. Shcherbakova, M. F. Balandrin, J. Fox, A. Ghatak, W. L. Heaton and R. L. Conklin, *Bioorg. Med. Chem. Lett.*, 15, 1557 (2005).
- 8. W. Lietz and H. Matties, Acta Biol. Med. Ger., 13, 591 (1964).
- 9. B. Das and J. Banerjee, Chemistry Lett., 33, 960 (2004).
- 10. Z. D. Jiang and R. F. Chen, Synth. Commun., 35, 503 (2005).
- 11. L. M. Wang, J. J. Xia, F. Qin, C. T. Qian and J. Sun, Synthesis, 1241 (2003).
- A. R. Khosropour, I. Mohammadpoor-Baltork and H. Ghorbankhani, *Tetrahedron Lett.*, 47, 3561 (2006).
- M. Narasimhulu, K. C. Mahesh, T. S. Reddy, K. Rajesh and Y. Venkateswarlu, *Tetrahedron Lett.*, 47, 4381 (2006).
- K. Ighilahriz, B. Boutemeur, F. Chami, C. Rabia, M. Hamdi and S. M. Hamdi, *Molecules*, 13, 779 (2008).
- 15. H. A. Oskooie, B. Baghernezhad and M. M. Heravi, Indian J. Heterocycl. Chem., 17, 95 (2007).

- 16. M. A. Chari and D. S. K. Mukkanti, Catal. Commun., 7, 787 (2006).
- 17. M. Jereb, D. Vražič and M. Zupan, Tetrahedron Lett., 50, 2347 (2009).
- 18. X. Gang, Y. Zhu, H. Birch, H. A. Hjuler and N. J. Bjerrum, Appl. Catal, A: Gen., 261, 91 (2004).
- 19. P. Phukan, Tetrahedron Lett., 45, 4785 (2004).
- M. A. Zolfigol, P. Salehi, A. Khorramabadi-Zad and M. Shayegh, J. Mol. Catal. A: Chem., 261, 88 (2007).
- 21. K. V. N. S. Srinivas and B. Das, Synthesis, 2091 (2004).
- 22. T. Ando, D. Kano, S. Minakata, I. Ryu and M. Komatsu, Tetrahedron, 54, 13485 (1998).
- 23. J. S. Yadav, B. V. S. Reddy, K. V. Rao, K. S. Raj, P. P. Rao, A. R. Prasad and D. Gunasekar, *Tetrahedron Lett.*, **45**, 6505 (2004).
- 24. C. M. Chu, S. Gao, M. N. V. Sastry and C. F. Yao, Tetrahedron Lett., 46, 4971 (2005).
- 25. B. Das, A. Majhi, K. R. Reddy and K. Venkateswarlu, J. Mol. Catal. A: Chem., 263, 273 (2007).
- 26. S. P. Chavan, P. Sharma, R. Krishna and M. Thakkar, Tetrahedron Lett., 44, 3001 (2003).